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Lead forms insoluble $PbSO_4$, $PbSO_3$, $PbCO_3$, PbS , PbI_2 , $PbCl_2$. PbS is a black precipitate,. PbI_2 is a yellow precipitate.. All the others are white precipitates. If a Lead(II)nitrate(V) solution is added to a substance/ solution/ compound : A yellow ppt shows presence of I-ions; A black ppt shows presence of S²⁻ ions; A white ppt shows presence of SO_4^{2-} , SO_3^{2-} , CO_3^{2-} -Cl-; If the white ...

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